

Final Exam - Thurs Dec 15 9:15 AM G-6

### Chapter 1

bonding + anti-bonding

$\sigma$  +  $\pi$  bonds

molecular geometry

hybridization  $\rightarrow$  CH<sub>4</sub>

acids + bases - hybridization, inductive, resonance

(pKa)

### Chapter 2

Nomenclature - Common

- alkanes, alkenes, alkynes, alcohols,
- alkyl halides, alkenols; no ethers, amines
- cyclic compounds

Rotomers

- staggered, 'eclipsed', syn, anti, gauche
- line structures, Newman projections

Rings + angle strain

Cyclohexane - ~~eclipsed, chair~~ equatorial, axial  
- chair, boat, ring flips

### Chapter 3

E vs Z; cis vs Trans

Kinetics, thermodynamics, equilibrium

$\hookrightarrow S_{w1} \text{ vs } S_{w2}$  (RLS)  $\rightarrow$  RCD, Hess's law

Chapter 4

Markovnikov addition  
List of alkene rxns

Chapter 5

## Stereochemistry

R or S? phantom atoms

enantiomer, diastereomer, epimer

optical rotation

racemic

Chapter 6<sup>MESO</sup>

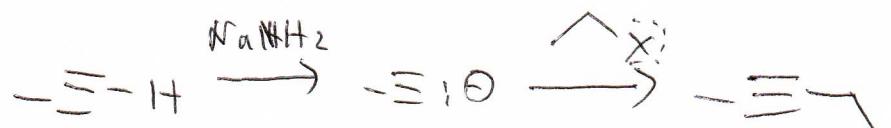
## Alkynes

Hydrogenations (full, cis, trans)

Hydroboration ( $\text{Ti}_2\text{BH}$ )

Tautomerization (base + acid)

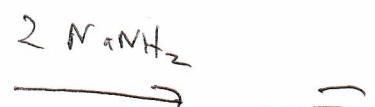
Alkylation



Tautomerization



(acid)

Chapter 7

resonance + delocalization  $\rightarrow$  conjugation  
smog

Chapters 8+9

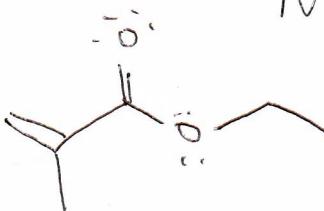
S<sub>N</sub>1, S<sub>N</sub>2, E1, E2

substrate, nucleophile, leaving group, solvent

Chapter 12

free radical halogenation

End of Chem 12A



NMR

ethyl methacrylate