

Nomenclature -

- alcohols, alkoxides, alkenes, aldehydes, ketones, benzene

Alcohols

Leaving groups - tosyl chloride, SOCl_2 ; PBr_3

Hydration; Dehydration

Oxidation; PCC vs. $\text{CrO}_3/\text{H}_2\text{O}$

Alkoxide; $\text{Na} + \text{NaH} \rightarrow$ Williamson Ether synthesis

Aldehydes + Ketones

Reactivity of aldehydes vs ketones

POAD - hydrates; acetals + ketals; imine + enamines;

cyanohydrins, oximes, hydrazones

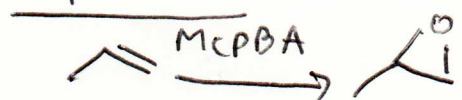
Reduction - NaBH_4 , LiAlH_4

Alkylation - Grignard; Wittig

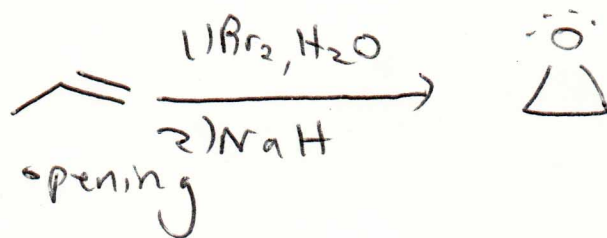
Wolff-Kishner reduction (hydrazones)

Protecting Groups

DHP; TBDMSCl

Epoxides

cationic vs anionic



opening

conjugation

SMOG

NO Kinetic vs Thermodynamics

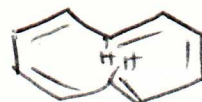
MO description of allyl C=C + C=C

Conjugated dienes

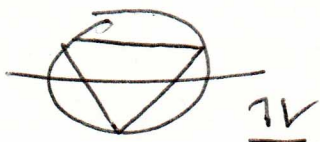
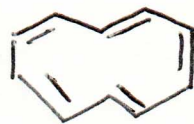
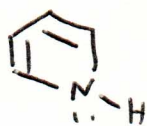
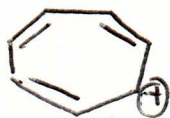
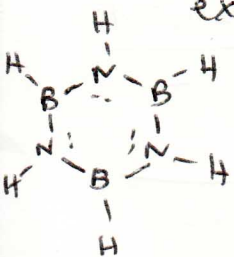
bonding, non-bonding, anti-bonding

Aromaticity

is it; aromatic, non-aromatic, anti-aromatic



MO of cyclobutadiene, benzene examples!



Synthesis: nitration, sulfonation, alkylation, acylation, halogenation

Pericyclic reactions

HOMO/LUMO theory

ground vs excited states

conrotatory / disrotatory cyclizations

Diels-Alder - s-cis vs s-trans;

endo + exo + secondary orbital effects.

End of Chem 12B

