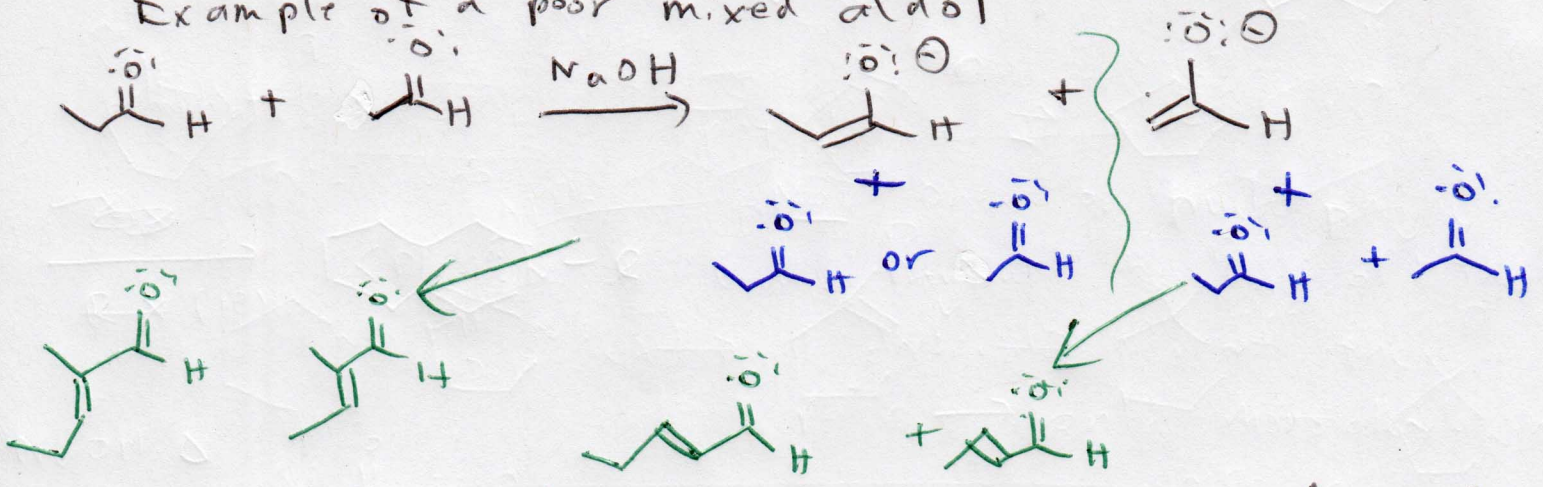


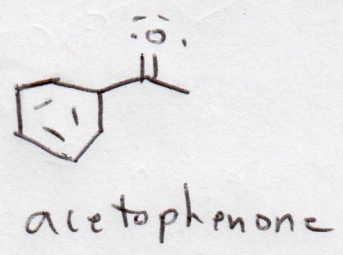
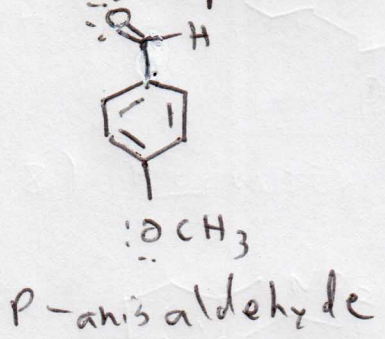
1) Aldol condensation

Mixed aldol

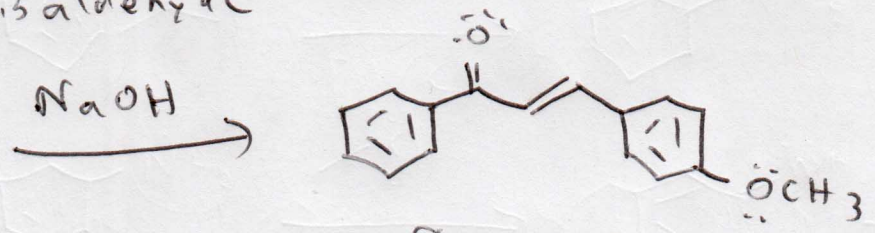
Example of a poor mixed aldol



Example of an ideal crossed-aldol condensation



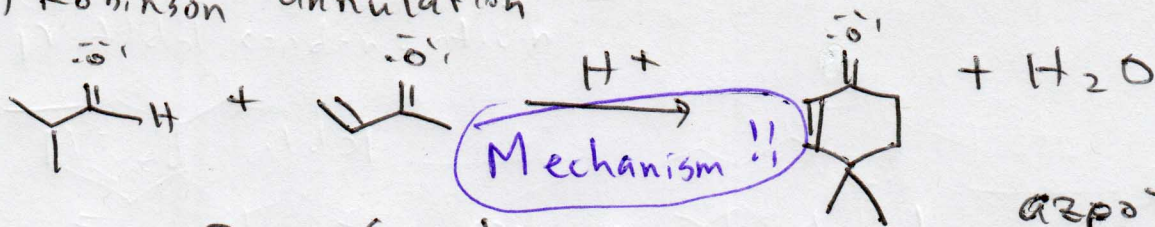
- Only one reagent is able to form an enolate
- Aldehyde much more reactive than ketone



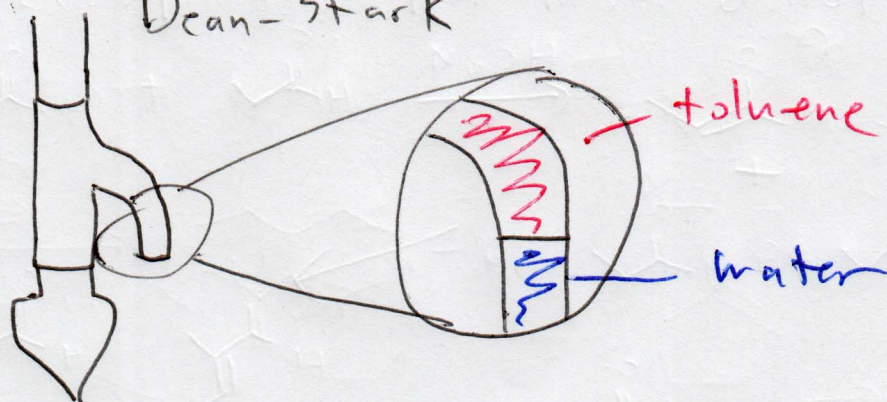
Spectroscopic confirmation

- The product contains a conjugate C=O that would have a noticeably different frequency than either starting material.
- The product contains a conjugate alkene not present in either reagent.

2) Robinson annulation

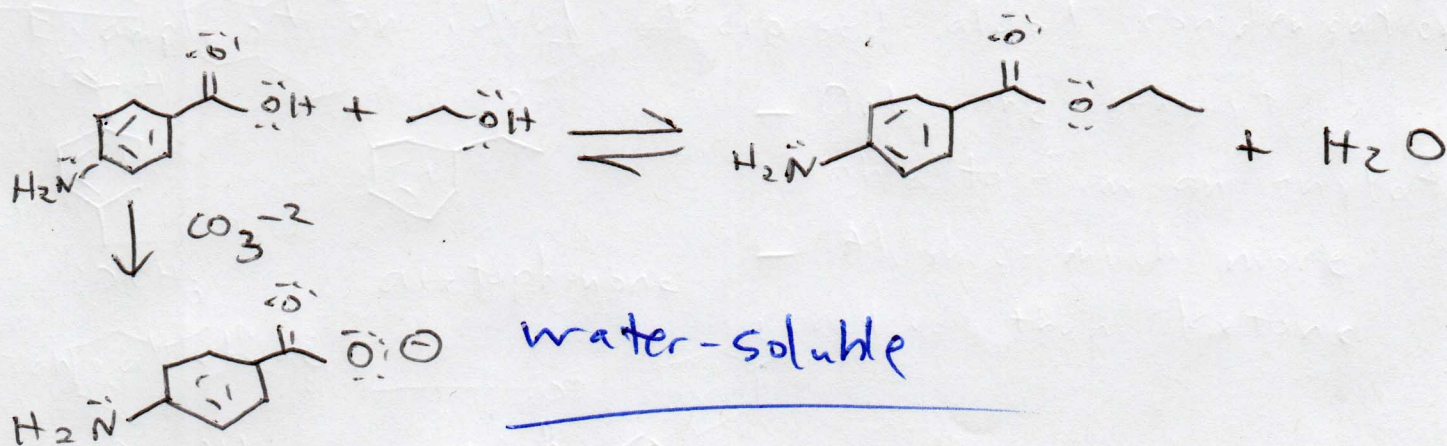


Dean-Stark

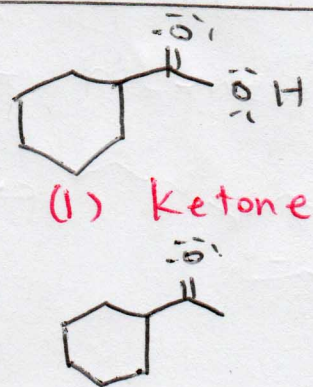
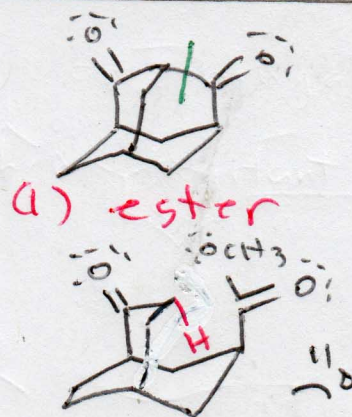
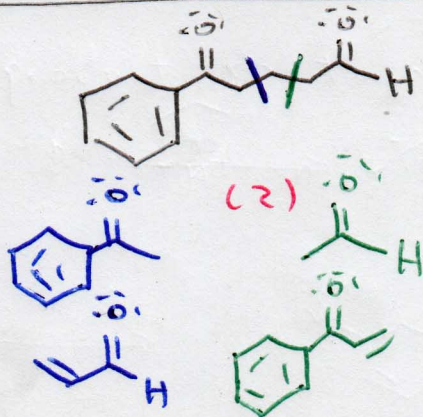
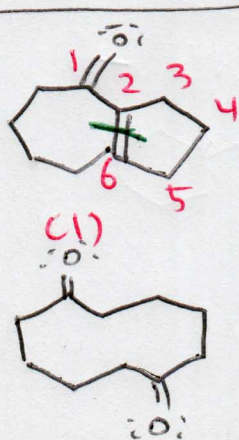


azeotrope - a gaseous mixture that phase separates upon condensation

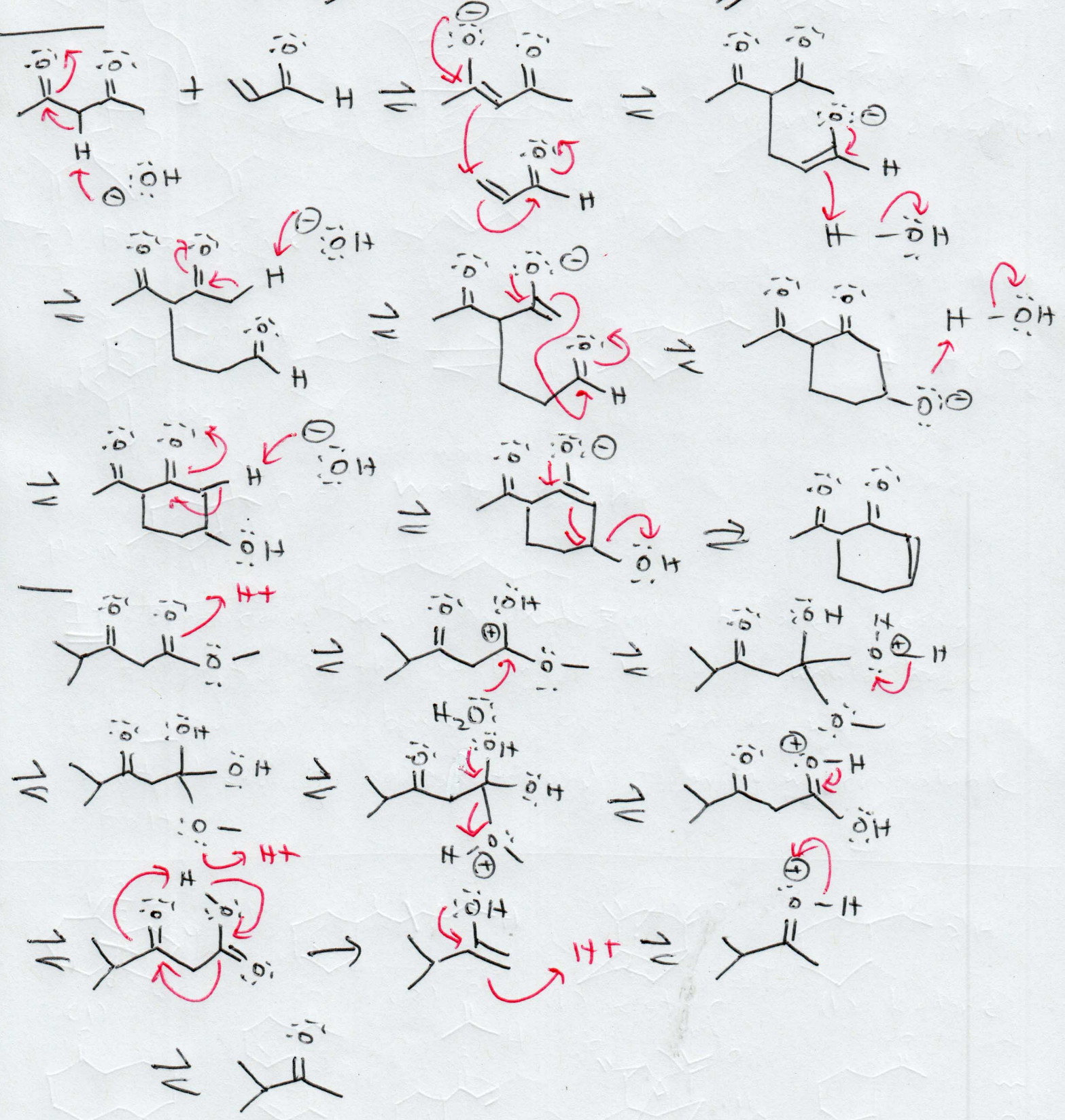
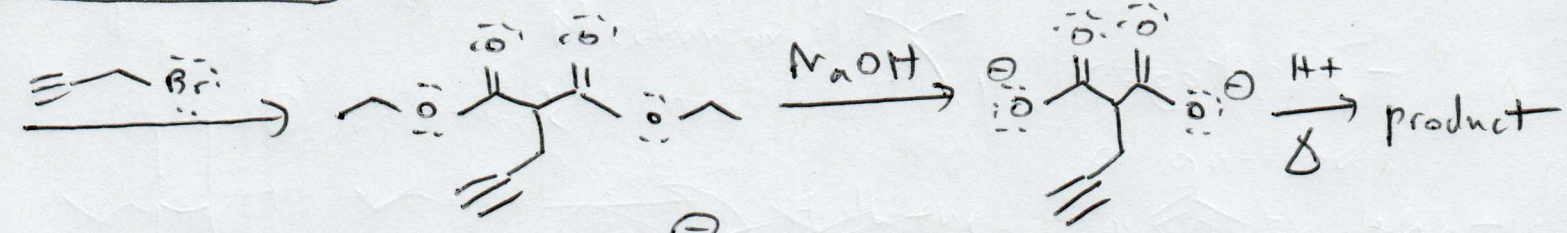
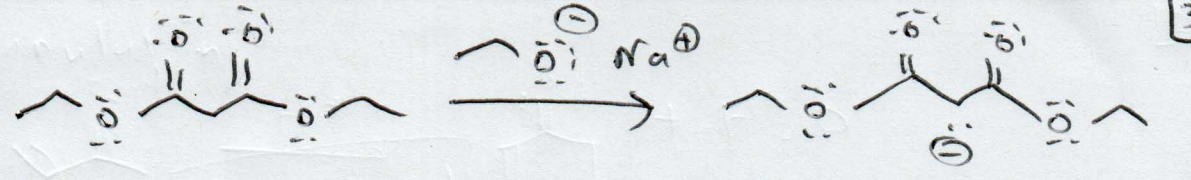
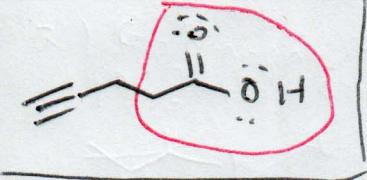
3) Benzocaine

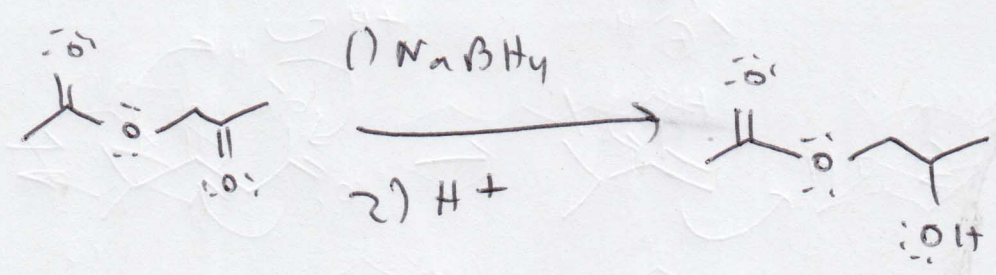
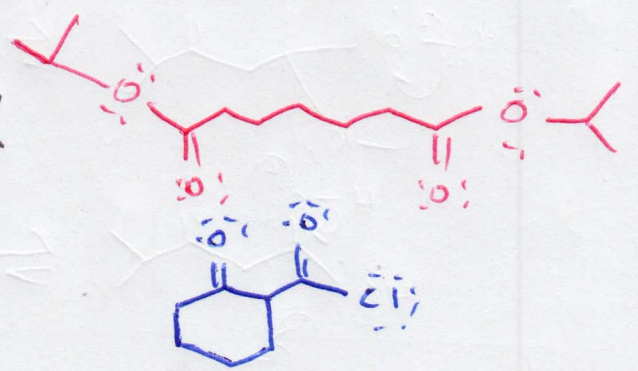
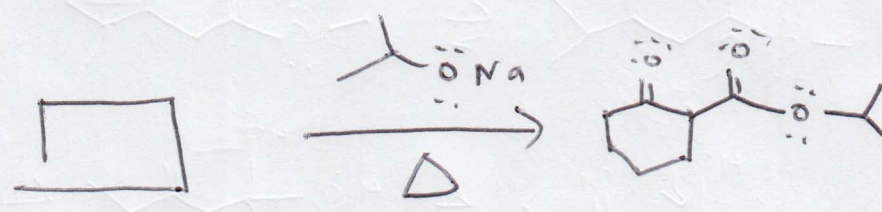
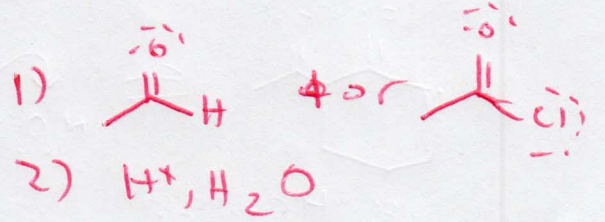
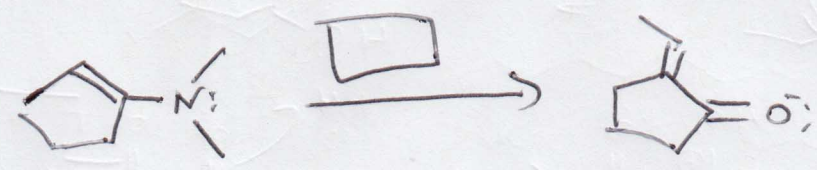
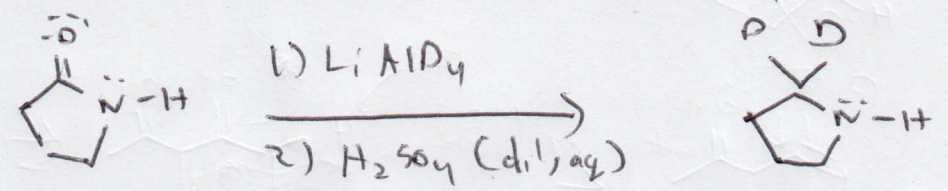
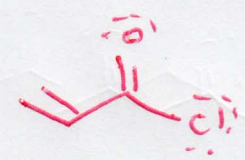
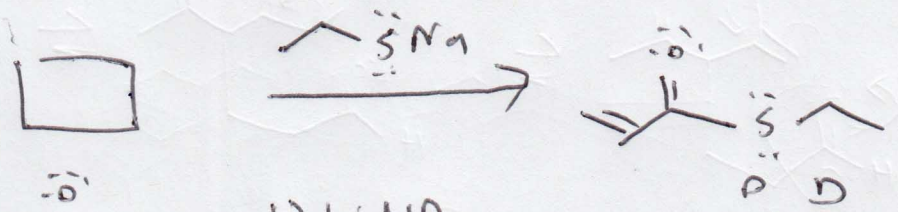
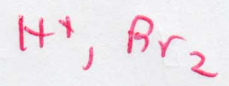
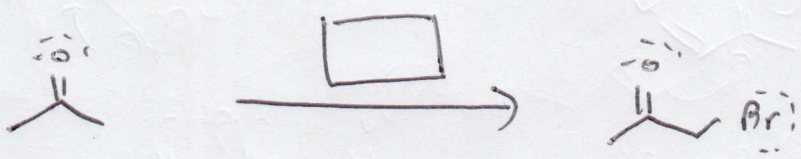
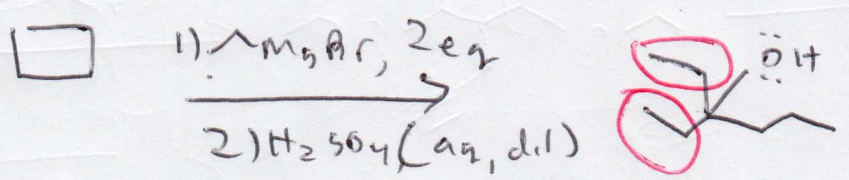
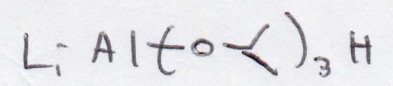
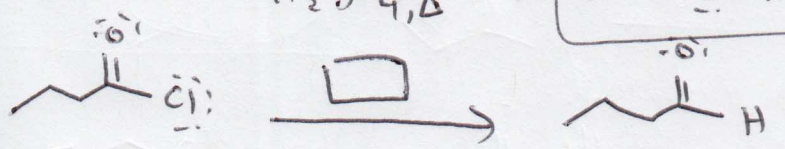
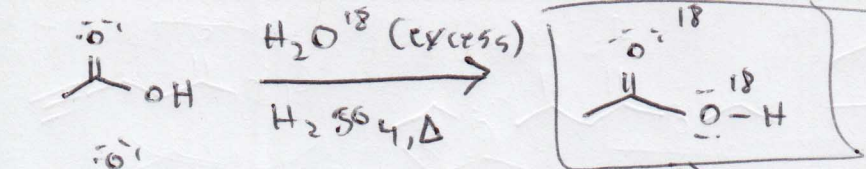


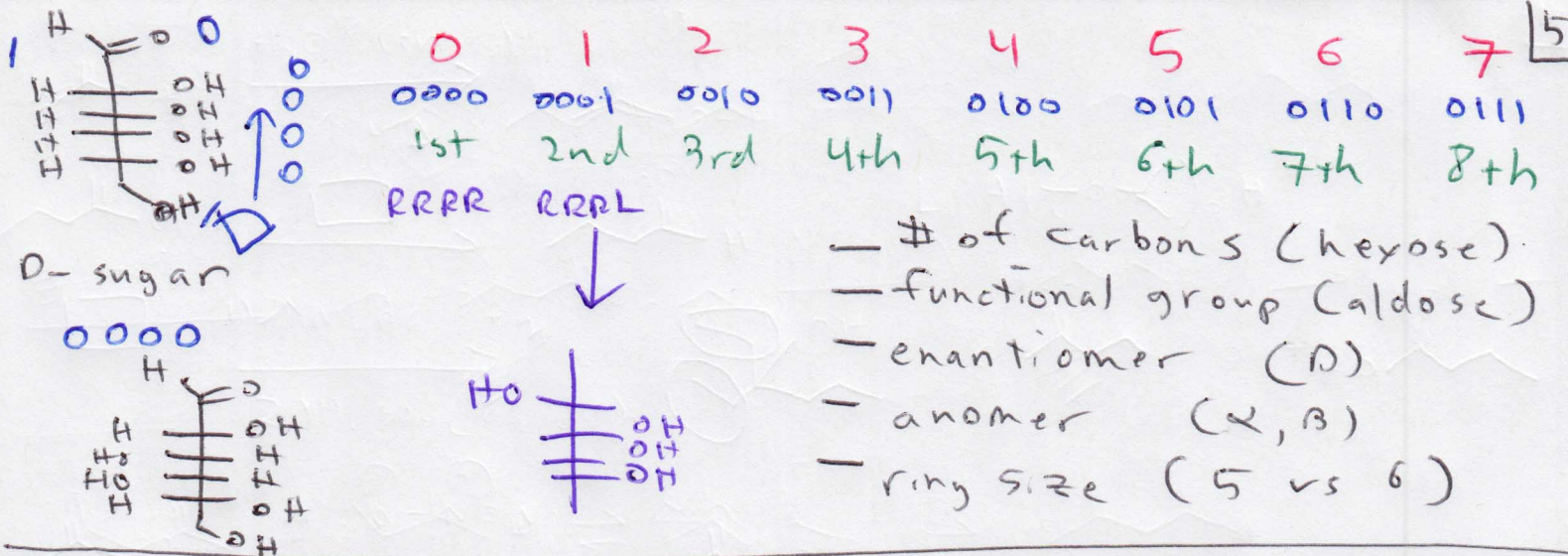
If benzocaine was hydrolyzed before being subjected to the carbonate wash, it would have reformed a carboxylic acid, which would be neutralized by carbonate and form a water-soluble salt.



!!!

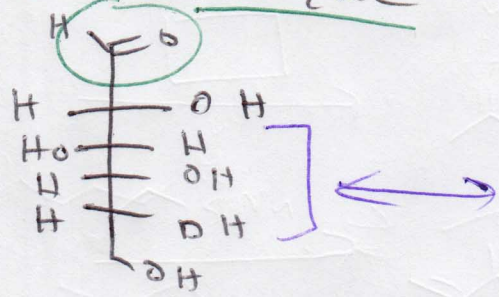




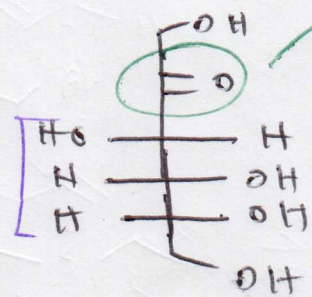


aldose - aldehyde

ketose - ketone



D-glucose



D-fructose

hexose - 6 carbons

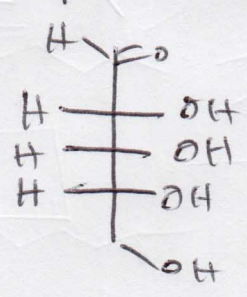
aldose + hexose → aldohexose

pentose - 5 carbons

tetraose - 4 carbons

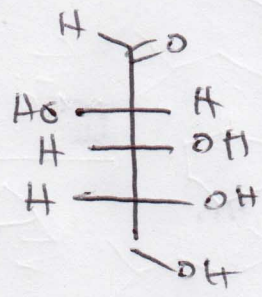
triose - 3 carbons

aldopentoses



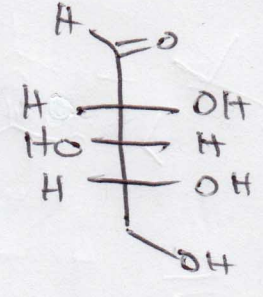
0000

D-ribose



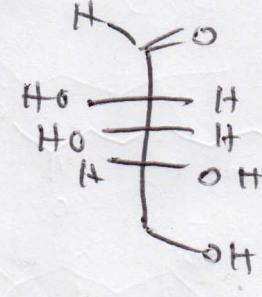
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D-arabinose



0010

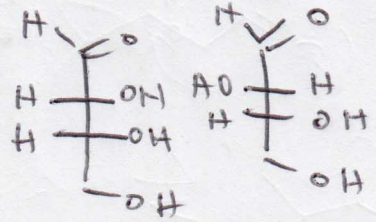
D-xylose



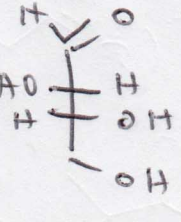
0011

D-lyxose

aldotetraose

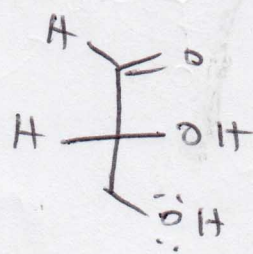


D-erythrose

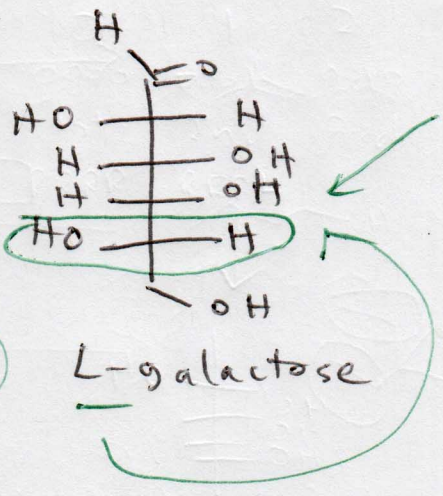
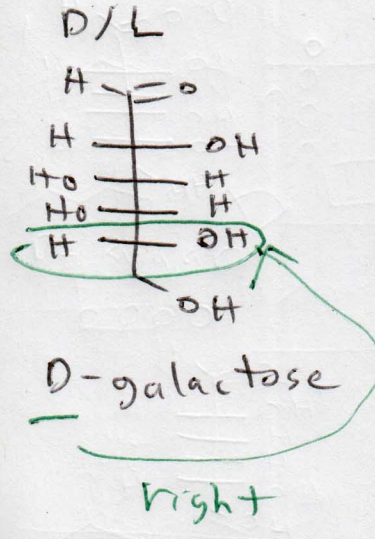


D-threose

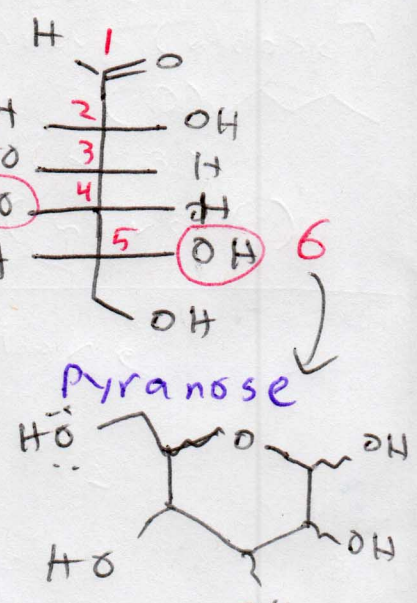
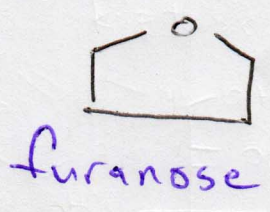
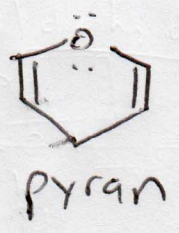
aldotriose



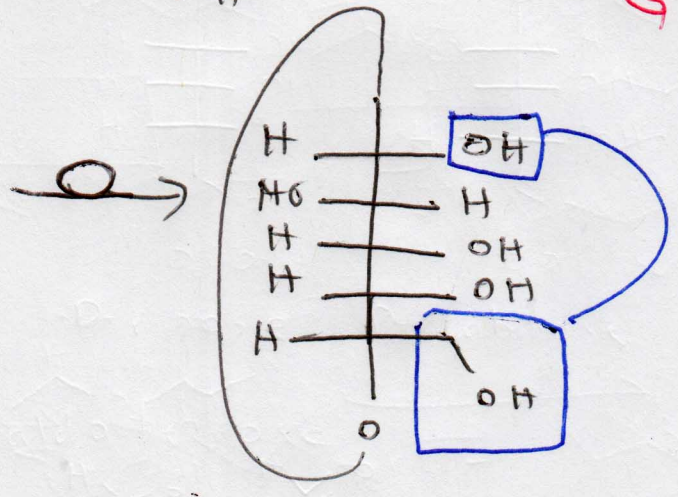
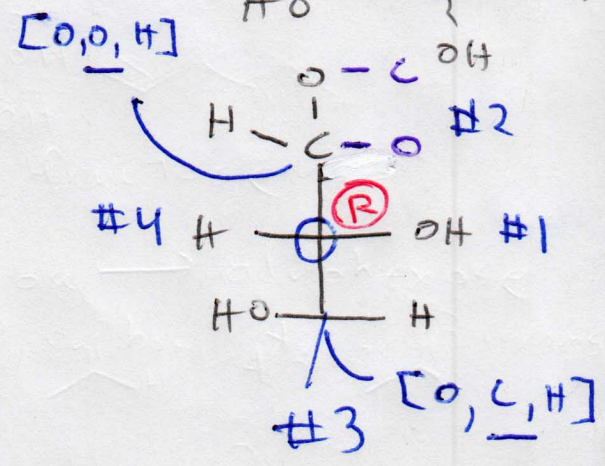
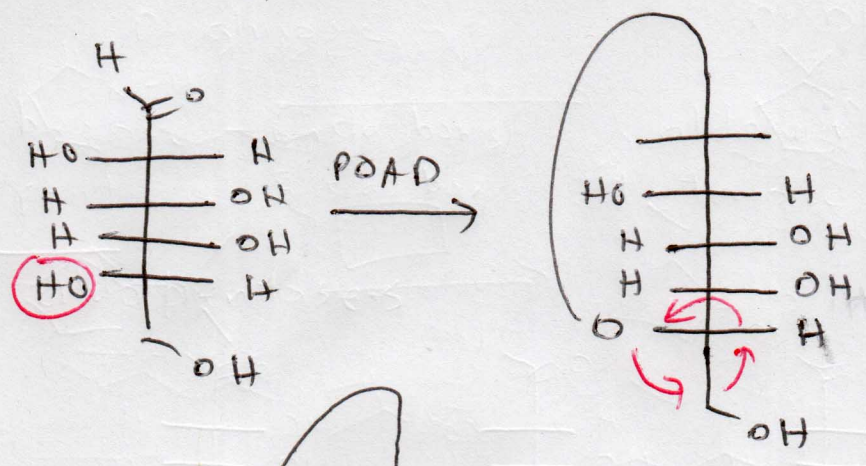
D-glyceraldehyde



D/L determined by last stereocenter



β -L-galactopyranose



β -
cis

Haworth

