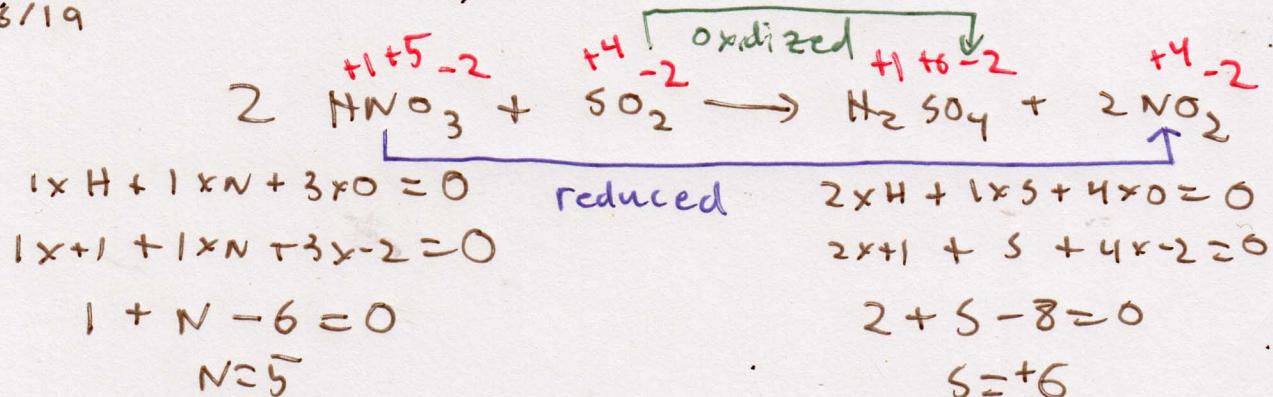
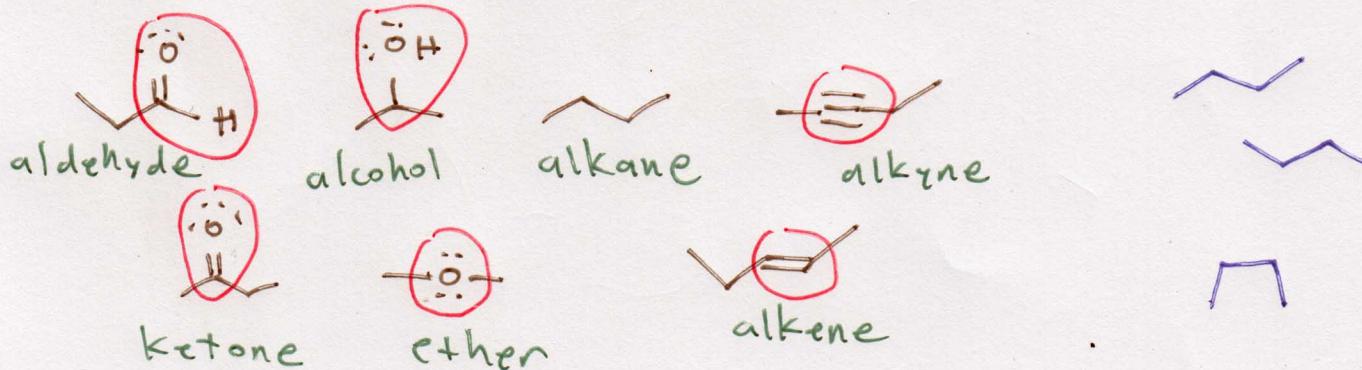
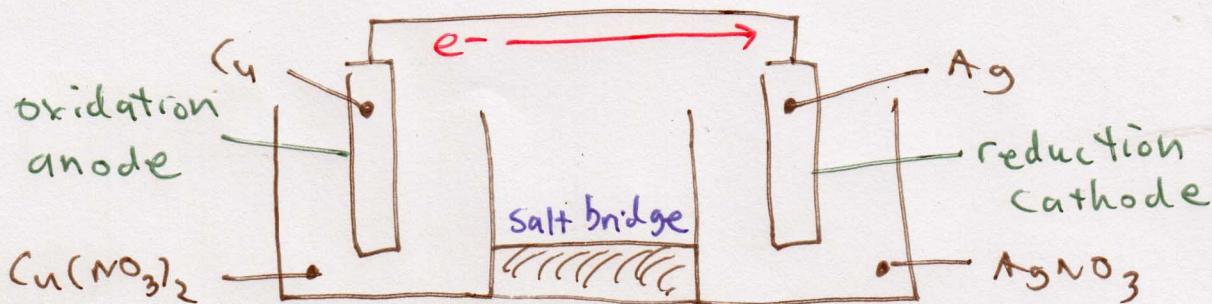


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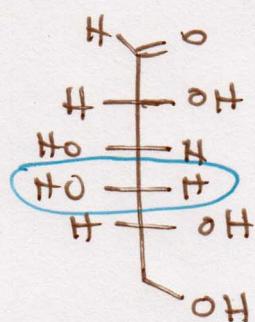
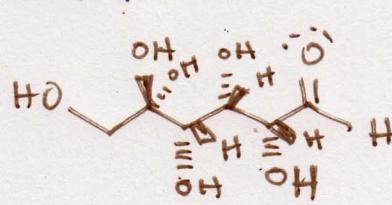
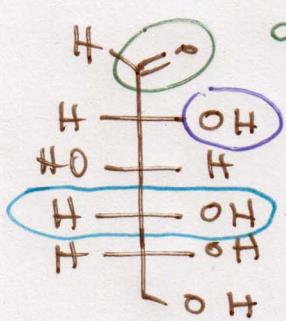
HNO_3 was reduced, so it was the oxidizing agent

SO_2 was oxidized, so it was the reducing agent



Carbohydrates — $(\text{CH}_2\text{O})_x$

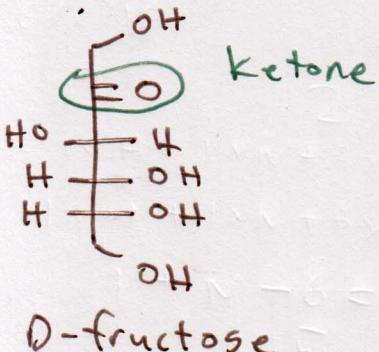
aldose — a sugar that contains an aldehyde



D-glucose

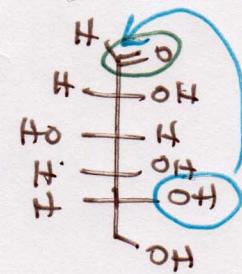
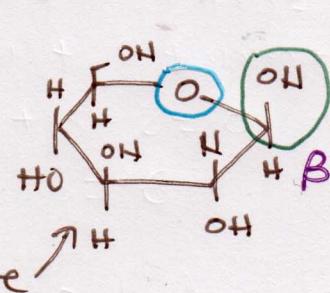
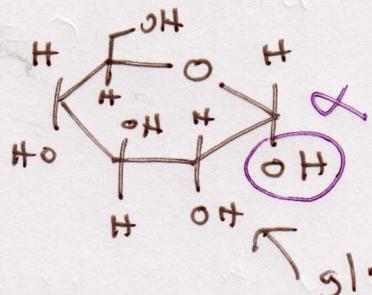
glucose + galactose are stereoisomers
stereocenter — an atom that causes stereoisomers to form depending how the geometry of the atom is configured

D-galactose



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Ketose - a sugar that contains a ketone
glucose & fructose are constitutional isomers



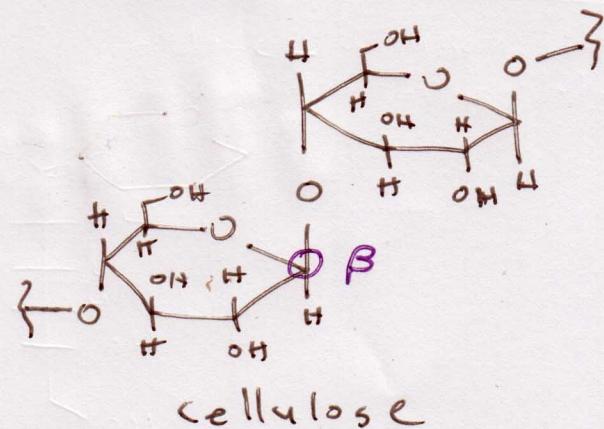
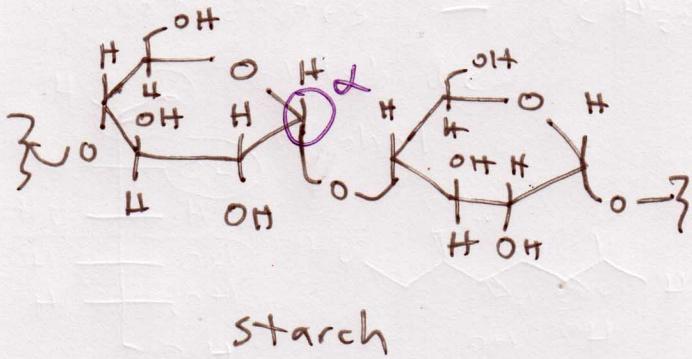
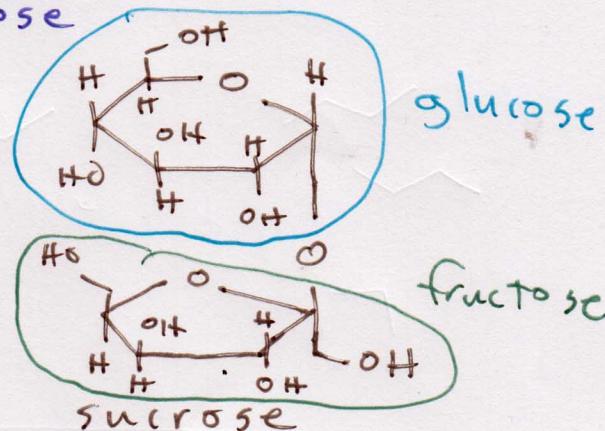
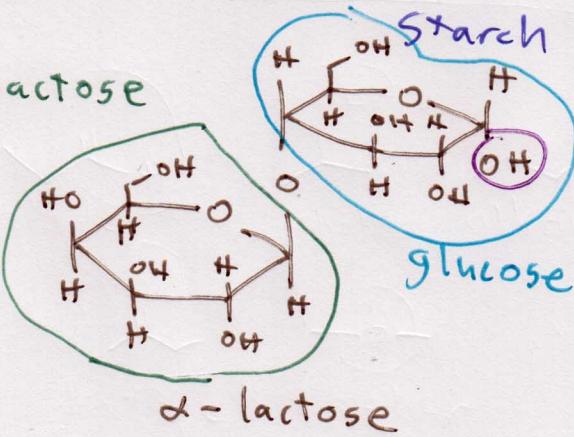
Monosaccharide - a single sugar unit that cannot be broken apart by reaction with water

glucose, fructose, galactose, ribose

disaccharide - a sugar composed of two individual units
sucrose, lactose, maltose

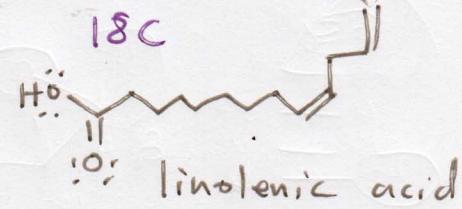
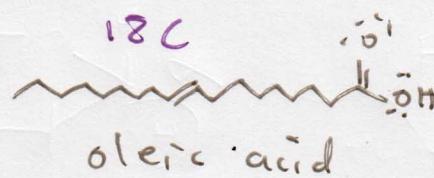
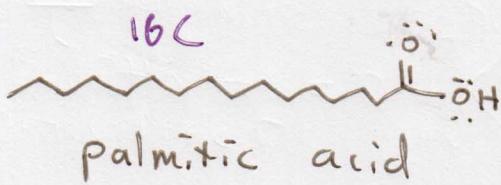
polysaccharide - a sugar composed of many individual units

galactose



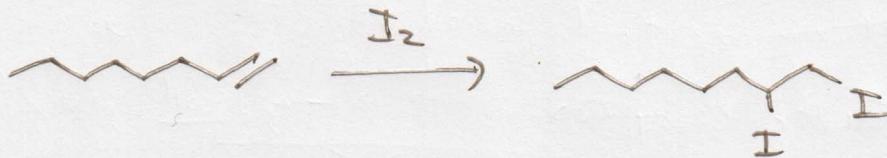
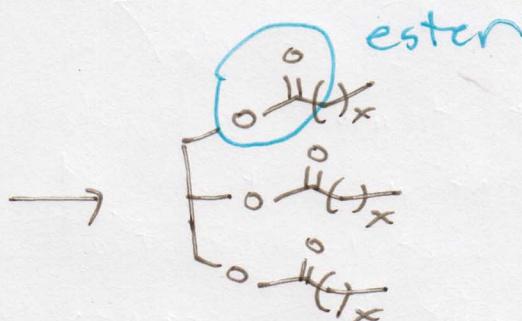
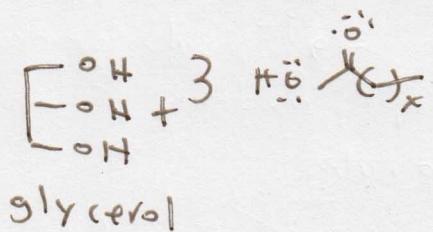
Lipid - a biological molecule that is soluble in non-polar materials (fats) (Q3)

fatty acids - long-chain carboxylic acids

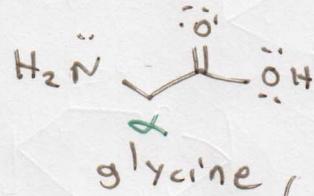


1,2,3

fat - triglyceride

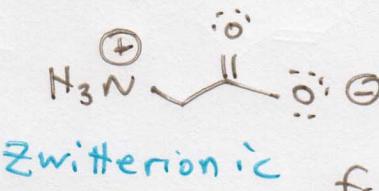


amino acids

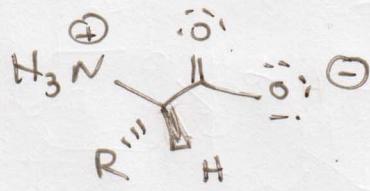


has both positive and negative charge

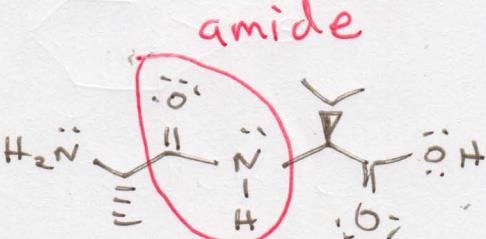
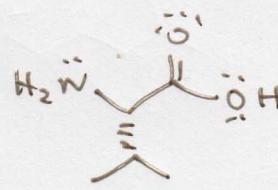
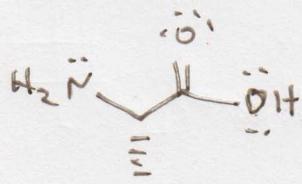
α -amino acid - an amino acid in which the amino group is attached to the carbon immediately next to the carboxylic acid



Since amino acids contain both an acidic and basic functional group, amino acids frequently exist in charged form.



R = radical (the organic equivalent of a variable)



dipeptide - a combination of two amino acids