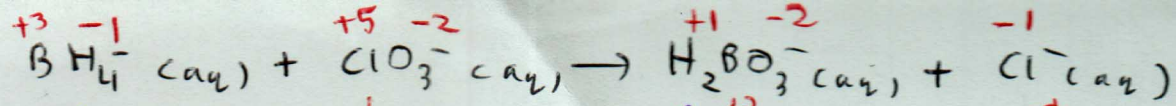


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$\text{BH}_4^-$  oxidized  
 reducing agent  
 $\text{ClO}_3^-$  reduced  
 oxidizing agent

H loses  $2 e^- \rightarrow$  oxidation each  
 Cl gains  $6 e^-$  each  $\rightarrow$  reduced

